ps of all the

5

10

15

20

25

Figure 3 is a table of 741 calcium channel antagonists according to the present invention.

## **DETAILED DESCRIPTION OF THE INVENTION**

Biological systems in general are controlled by molecular interactions between bioactive ligands and their receptors, in which the receptor "recognizes" a molecule or a portion thereof (i.e., a ligand domain) to produce a biological effect. The voltage-gated Ca<sup>++</sup> channels are considered to be pharmacological receptors: they possess specific binding sites for ligands having agonist and antagonist activities; the binding of ligands to such sites allosterically modulates Ca<sup>++</sup> flux through the channel; the channel properties (i.e., gating and ion selectivity) are regulatable; and various channels are known to associate with G-proteins (D. Rampe and D.J. Triggle, *Prog. Drug Res.* 40: 191-238 (1993). Accordingly, diseases or conditions that involve, or are mediated by, Ca<sup>++</sup> channels can be treated with pharmacologically active ligands that interact with such channels to initiate, modulate or abrogate transporter activity.

The interaction of a Ca<sup>++</sup> channel and a Ca<sup>++</sup> channel-binding ligand may be described in terms of "affinity" and "specificity". The "affinity" and "specificity" of any given ligand-Ca<sup>++</sup> channel interaction is dependent upon the complementarity of molecular binding surfaces and the energetic costs of complexation (*i.e.*, the net difference in free energy ΔG between bound and free states). Affinity may be quantified by the equilibrium constant of complex formation, the ratio of on/off rate constants, and/or by the free energy of complex formation. Specificity relates to the difference in binding affinity of a ligand for different receptors.

The net free energy of interaction of such ligands with a Ca<sup>→</sup> channel is the difference between energetic gains (enthalpy gained through molecular complementarity and entropy gained through the hydrophobic